Journal of Holography Applications in Physics Volume 1, Issue 1, Autumn 2021, 71-83 © Available online at http://jhap.du.ac.ir DOI:10.22128/jhap.2021.453.1003

Journal of Holography Applications in Physics

### Regular article

# Holographic application in cosmology: Thermodynamics of the Van der Waals cosmic fluid

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Received: August 28, 2021; Revised: October 3, 2021; Accepted: October 10, 2021.

**Abstract**. This paper is devoted to investigate the thermodynamic stability of a generic cosmological fluid known as Van der Waals fluid in the context of flat FRW universe. It is treated as a perfect fluid that obeys the equation of state

$$P = \frac{\gamma \rho}{1 - \beta \rho} - \alpha \rho^2, \quad 0 \le \gamma < 1$$

where  $\rho$  stands for energy density and P stands for pressure of the fluid. In this regard, we discuss the behavior of physical parameters to analyze the evolution of the universe. We investigate whether the cosmological scenario fulfills the third law of thermodynamics using specific heat formalism. Next we discuss the thermal equation of state and by means of adiabatic, specific heat and isothermal conditions from classical thermodynamics we examine the thermal stability.

Keywords: Thermodynamics; gas; stability.

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### 1 Introduction

Cosmological observations such as type Ia supernova [1, 2, 3, 4], the large scale structure [5, 6, 7, 8], cosmic microwave background (CMB) anisotropies [9, 10, 11] indicate that the universe endures an accelerating expansion. The source of this phenomenon is suspected to be an exotic fluid that violates strong energy condition and possesses a large negative pressure, dubbed as dark energy. According to Plancks observational data [12] about 68.3 percent of the total cosmic budget is occupied by Dark energy, while about 26.8 percent is filled with dark matter and about 4.9 percent is usual baryonic matter. There have been a prolonged attempt to reconcile the physical nature of dark energy. An overwhelming flood of dynamical dark energy models such as quintessence, tachyon [13], ghost [14], k-essence [15], fermionic field [16, 17, 18], phantom [19], Chaplygin gas [20], holographic dark energy (HDE) [21, 22, 23], new agegraphic dark energy (NADE) [24, 25] and modified gravity models such as f(R) gravity [26, 27], f(T) gravity [28, 29, 30], Hořava-Lifshitz gravity [31, 32, 33, 34] have been proposed in literature.

Like Chaplygin gas family, another single-component fluid known as Van der Waals fluid [35] has attracted much attention for unification of different fluid. The main feature of this model is the ability to reproduce the accelerated and matter-dominated phases with a single component. A simple description by a perfect fluid equation of state is not realistic in describing the all the phases of evolution of the universe. The properties of Van der Waals fluid have been analyzed in [36]. It is indeed the holographic description of the dark energy, so by having information about the Van der Waals fluid we can obtain knowledge about the accelerating expansion of universe. In [37, 38] a mixture of two fluids, taking the Van der Waals fluid as dark energy and the perfect gas equation of state for the matter, have been considered to get a whole dynamics of the universe. In Ref. [39], a toy model of the Universe is considered with generalized ghost dark energy, Van der Waals gas and a phenomenologically modified fluid. They have studied the unusual connection among different fluids. Although there have been a lot of work [40, 41, 42] discussing various aspects of Van der Waals fluid in order to reconcile the standard cosmological model with observations, there has not been done a full analysis of the thermodynamic constraints of such a fluid. Thermodynamics of Chaplygin gas model has been studied by Myung [43]. Santos et al [44, 45] have studied the thermodynamic stability of the generalized and modified Chaplygin gas models. Thermodynamics of Modified Chaplygin Gas and Tachyonic Field have been analyzed in ref. [46]. Thermodynamic stability of generalized cosmic Chaplygin gas has been studied by Sharif et al [47]. Motivated by these works, here, we examine the thermodynamic stability of Van der Waals fluid in the background of a flat FRW universe.

The paper is organized as follows. In section 2, we study the behavior of physical parameters such as pressure, EoS as well as deceleration parameters and analyze the stability using square speed of sound. Section 3 deals with the thermodynamic stability of Van der Waals fluid. We devote the last section for summarization of the results.

### $\mathbf{2}$ Physical Features of Van der Waals Fluid

Here we assume the flat Friedmann-Robertson-Walker (FRW) model of the universe represented by the following line element:

$$ds^{2} = -dt^{2} + a^{2}(t)(dr^{2} + r^{2}d\theta^{2} + r^{2}\sin^{2}\theta d\phi^{2}), \tag{1}$$

where a(t) is the scale factor. Now we assume the equation of state of Van der Waals fluid as [35]

$$P = \frac{\gamma \rho}{1 - \beta \rho} - \alpha \rho^2, 0 \le \gamma < 1, \alpha = 3p_c \rho_c^{-2}, \beta = (3\rho_c)^{-1},$$
 (2)

where  $\rho$  and P are respectively the energy density and pressure of the fluid. Here  $\rho_c$  and  $p_c$  are the density and the pressure of the cosmic fluid at the Van der Waals critical point. The above equation reduces to the perfect fluid case in the limit  $\alpha, \beta \to 0$ . The energy density of the fluid can be written in the form:

$$\rho = \frac{U}{V},\tag{3}$$

where U is the internal energy and V is the volume. From classical thermodynamics, the relation between U, V and P can be written in the form [47, 48]

$$\frac{dU}{dV} = -P. (4)$$

From equations (2) - (4), we get the following first order ordinary differential equation

$$\frac{dU}{dV} + \frac{\gamma \frac{U}{V}}{1 - \beta \frac{U}{V}} = \alpha \left(\frac{U}{V}\right)^2. \tag{5}$$

Assuming the binomial expansion upto first order, we obtain the approximate solution

$$U \approx V \left[ \frac{(\alpha + \beta) + \sqrt{(\alpha + \beta)^2 - 4(1 + \gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}} \right], \tag{6}$$

where k is an integration constant  $(\neq 0)$  which is either universal constant or a function of entropy (S). The above solution provides the solution of energy density as

$$\rho = \left\lceil \frac{(\alpha + \beta) + \sqrt{(\alpha + \beta)^2 - 4(1 + \gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}} \right\rceil$$
(7)

For small volumes  $(V \approx 0)$ , the energy density of the Van der Waals fluid behaves like

$$\rho = \left\lceil \frac{(\alpha + \beta) + \sqrt{(\alpha + \beta)^2 - 4(1 + \gamma)\alpha\beta}}{2\alpha\beta} \right\rceil$$
 (8)

with the condition  $(\alpha+\beta)^2 \geq 4(1+\gamma)\alpha\beta$  and hence the minimum value of  $\rho$  is  $\rho_{min} = \sqrt{\frac{1+\gamma}{\alpha\beta}}$ . Now we will discuss different physical parameters of the model.

### 2.1 Pressure

From equations (2) and (7), we obtain the expression pressure in terms of V as in the following form:

$$P = \frac{\gamma[(\alpha+\beta) + \sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}]}{\alpha\beta - [\beta^2 + 2(\frac{V}{k})^{2(1+\gamma)} + \beta\sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}]}$$

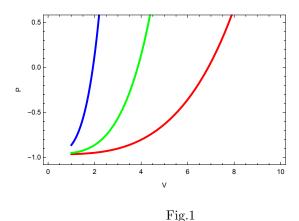


Figure 1: Plots of P versus V for  $\gamma = 0.7, \beta = 1, \alpha = 20$  and k = 5 (blue curve), k = 10(red curve) and k = 20 (green curve).

$$-\alpha \left[ \frac{(\alpha+\beta) + \sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]} \right]^2$$
(9)

The trajectory of pressure given by the above equation against volume is drawn in figure 1 for different values of k = 5, 10, 20. Figure shows the positive and negative behavior of pressure. It is observed that the accelerating universe at small volume tends to decelerated phase of universe at large volume.

#### 2.2 EoS Parameter

From equations (2) and (7), we obtain the equation of state parameter in terms of V as in the following form:

$$\omega = \frac{P}{\rho} = \frac{2\gamma[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]}{\alpha\beta - [\beta^2 + 2(\frac{V}{k})^{2(1+\gamma)} + \beta\sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}]}$$

$$-\alpha \left[ \frac{(\alpha+\beta) + \sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]} \right]$$

$$= \begin{cases} \gamma, & V \gg k \\ -1, & V \ll k. \end{cases}$$
(10)

The EOS parameter  $\omega$  is drawn in figure 2 for different values of k = 5, 10, 20. The EOS parameter transits from -1 to positive values as volume increases. That means it yields cosmological constant model for small volume, then it generates the quintessence region and goes to positive region (tends to  $\gamma$ ).

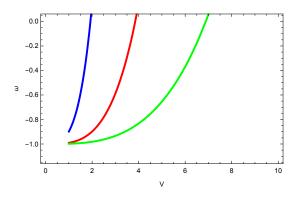


Fig.2

**Figure 2:** Plots of  $\omega$  versus V for  $\gamma = 0.7, \beta = 1, \alpha = 20$  k = 5 (blue curve), k = 10 (red curve) and k = 20 (green curve).

## 2.3 Deceleration Parameter

The deceleration parameter is given by

$$q = \frac{1}{2} + \frac{3P}{2\rho} = \frac{1}{2} + \frac{3\gamma[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]}{\alpha\beta - [\beta^2 + 2(\frac{V}{k})^{2(1+\gamma)} + \beta\sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}]}$$
$$-\alpha \left[ \frac{(\alpha+\beta) + \sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]} \right]$$
$$= \begin{cases} \frac{1}{2} + \frac{3}{2}\gamma, & V \gg k \\ -1, & V \ll k. \end{cases}$$
(11)

Figure 3 presents the trajectories of the deceleration parameter q against V for different values of k. From the graph we can see that q increases from -1 to the positive value (tends to  $\frac{1+3\gamma}{2}$ ). It describes acceleration at small volumes whereas at large volume, it exhibits decelerating behavior for different values of k. So, a transition from accelerating to decelerating universe is observed. It shows  $q \to -1$  with decreasing V i.e it yields cosmological constant model as  $V \to 1$ .

# 2.4 Square Speed of Sound

To discuss the classical stability of the model, we need to obtain the square speed of sound which is given by

$$\begin{split} V_s^2 &= \left(\frac{\partial P}{\partial \rho}\right)_S &= \frac{4\gamma \left[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\right]^2}{\left[\alpha\beta - \left\{\beta^2 + 2(\frac{V}{k})^{2(1+\gamma)} + \beta\sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}\right\}\right]^2} \\ &- 2\alpha \frac{(\alpha+\beta) + \sqrt{(\alpha+\beta)^2 - 4(1+\gamma)\{\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}\}}}{2[\alpha\beta - (\frac{V}{k})^{2(1+\gamma)}]} \end{split}$$

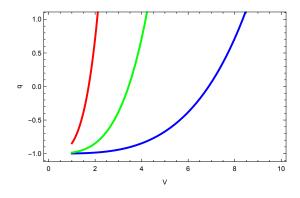


Fig.3

Figure 3:Plots of q versus V for  $\gamma = 0.7, \beta = 1, \alpha = 20$  k = 5 (blue curve), k = 10 (red curve) and k = 20 (green curve).

$$= \left\{ \begin{array}{ll} \gamma, & V \gg k \\ 4 \left[ \frac{\{\alpha(1+\gamma)-\beta\}\sqrt{(\alpha-\beta)^2 - 4\alpha\beta\gamma} + 3\alpha\beta\gamma - (\alpha-\beta)^2}}{\{\sqrt{(\alpha-\beta)^2 - 4\alpha\beta\gamma} - \alpha+\beta\}^2} \right], & V \ll k. \end{array} \right.$$
(12)

In figure 4 the squared speed of sound is plotted against V for different values of k. We observe that for  $0 \le V \lesssim 9$ , graph shows  $V_s^2 > 0$  while for  $V \gtrsim 9$ , graph shows  $V_s^2 < 0$ . So the model is classically stable for small volume and for large volume it shows unstable behavior.

### 3 Thermodynamic Stability of the Van der Waals Fluid

Now we will discuss the temperature behavior and the thermodynamic stability of the Van der Waals Fluid. To verify the thermodynamic stability conditions of this fluid along its evolution, it is necessary to determine (i) if the pressure is reduced through an adiabatic expansion  $\left(\frac{\partial P}{\partial V}\right)_S < 0$ , (ii) if the pressure reduces as the fluid expands at constant temperature T,  $\left(\frac{\partial P}{\partial V}\right)_T < 0$  and (iii) if the thermal capacity at constant volume,  $C_V > 0$ .

Differentiating equation (9) w.r.t. V, we obtain

$$\left(\frac{\partial P}{\partial V}\right)_{S} = \frac{(1+\gamma)\left(\frac{V}{k}\right)^{2(1+\gamma)}}{VX^{2}\sqrt{Y}} \left[ (\alpha+\beta)\sqrt{Y} + 2(1+\gamma)\left(\frac{V}{k}\right)^{2(1+\gamma)} - 2\gamma\alpha\beta + \beta^{2} + \alpha^{2} \right] \times \left[ 4\gamma X^{2} \left(\alpha\beta - (\beta^{2} + 2(\frac{V}{k})^{2(1+\gamma)} + \beta\sqrt{Y})\right)^{-2} - \frac{\alpha(\alpha+\beta+\sqrt{Y})}{X} \right], (13)$$

where  $X = \alpha \beta - (\frac{V}{k})^{2(1+\gamma)}$  and  $Y = (\alpha + \beta)^2 - 4(1+\gamma)X$ . When volume is very small, the above equation reduces to zero while for large volume, we have the following expression

$$\left(\frac{\partial P}{\partial V}\right)_{S} = \frac{-4(1+\gamma)\{(\beta-\alpha(1+\gamma))\sqrt{(\alpha-\beta)^{2}-4\alpha\beta\gamma}+(\alpha-\beta)^{2}-3\alpha\beta\gamma\}}{v\alpha^{2}\beta^{2}\{-\alpha+\beta+\sqrt{(\alpha-\beta)^{2}-4\alpha\beta\gamma}\}\sqrt{(\alpha-\beta)^{2}-4\alpha\beta\gamma}} \times \left[(\alpha+\beta)\sqrt{(\alpha-\beta)^{2}-4\alpha\beta\gamma}+\alpha^{2}+\beta^{2}-2\alpha\beta\gamma\right]. \tag{14}$$

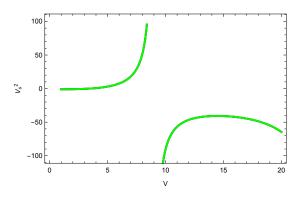


Fig.4

**Figure 4:**Plots of  $V_s^2$  versus V for  $\gamma = 0.7, \beta = 1, \alpha = 20$  k = 5 (blue curve), k = 10 (red curve) and k = 20 (green curve).

Figure 5(a) shows that  $\left(\frac{\partial P}{\partial V}\right)_S < 0$  at large volumes and at small volume, it is positive and tends to zero as  $V \to 0$ . So, the adiabatic condition is satisfied for all the considered values of k.

The specific heat capacity in constant volume is defined by

$$C_V = T \left(\frac{\partial S}{\partial T}\right)_V,\tag{15}$$

where the temperature can be obtained by the relation [47]

$$T = \frac{\partial U}{\partial S} = \left(\frac{\partial U}{\partial k}\right) \left(\frac{\partial k}{\partial S}\right) \tag{16}$$

Differentiating (6) with respect to k, we get

$$\frac{\partial U}{\partial k} = \frac{2(1+\gamma)(\frac{V}{k})^{2(1+\gamma)}}{kX\sqrt{Y}}[V(1+\gamma) - U\sqrt{Y}]$$
(17)

From equations (16) and (17), we obtain

$$T = \frac{2(1+\gamma)(\frac{V}{k})^{2(1+\gamma)}}{kX\sqrt{Y}}[V(1+\gamma) - U\sqrt{Y}]\left(\frac{\partial k}{\partial S}\right)$$
(18)

From equation (6), we can write (in the sense of dimensional analysis) [47]

$$[k]^{(1+\gamma)} = [U][V]^{\gamma}$$
 (19)

Using the relation [U] = [T][S], we obtain

$$[k] = [T]^{\frac{1}{(1+\gamma)}} [U]^{\frac{1}{(1+\gamma)}} [V]^{\frac{\gamma}{(1+\gamma)}}$$

$$(20)$$

From this result, we get

$$k = (\tau \nu^{\gamma} S)^{\frac{1}{(1+\gamma)}} \tag{21}$$

where  $\tau$  and  $\nu$  are constants having the dimensions of temperature and volume respectively. Differentiating (21), we obtain

$$\frac{\partial k}{\partial S} = \frac{1}{(1+\gamma)} \left( \frac{\tau \nu^{\gamma}}{S^{\gamma}} \right)^{\frac{1}{(1+\gamma)}} \tag{22}$$

Using (18) and (22) we have

$$T = \frac{2B^{\frac{1}{1+\gamma}}S^{-\frac{\gamma}{1+\gamma}}(\frac{V}{k})^{2(1+\gamma)}}{kX\sqrt{Y}}[V(1+\gamma) - U\sqrt{Y}]$$
(23)

where  $B = \tau \nu^{\gamma}$ . Using (6) and (21), the equation (23) becomes

$$T = -\frac{2BV^{3+2\gamma}}{X'^2\sqrt{Y'}}[(1+\gamma)X' + ((\alpha+\beta)BS + \sqrt{Y'})\sqrt{Y'}],$$
 (24)

where  $X' = B^2 S^2 X$  and  $Y' = B^2 S^2 Y$ . When T = 0, the entropy S = 0 which implies that third law of thermodynamics is satisfied for our Van der Waals fluid model. Differentiating eq.(24) with respect to S, we obtain

$$\frac{\partial T}{\partial S} = \frac{2V^{3+2\gamma}B^2}{X'^3Y'^{\frac{3}{2}}} [2BS\alpha\beta\{3X'Y'(1+\gamma) - 2X'^2(1+\gamma)^2 + 2Y'^2\} + Y'^{\frac{3}{2}}(\alpha+\beta)\{4\alpha\beta B^2S^2 - X'\} + BSX'(\alpha+\beta)^2\{(1+\gamma)X' + 2Y'\}]$$
(25)

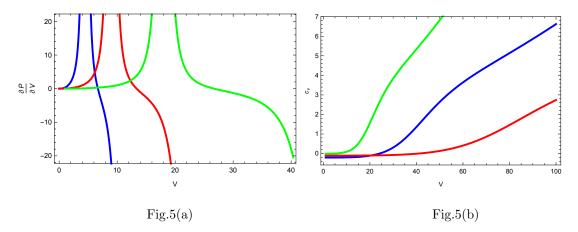
Now from equation (15), we have the expression of specific heat capacity as in the following form:

$$C_{V} = -[X'^{2}Y(1+\gamma) + X'Y'^{\frac{3}{2}}\{SB(\alpha+\beta) + \sqrt{Y}\}] \times \left(B[2BS\alpha\beta\{3X'Y'(1+\gamma) - 2X'^{2}(1+\gamma)^{2} + 2Y'^{2}\}\right) + Y'^{\frac{3}{2}}(\alpha+\beta)\{4\alpha\beta B^{2}S^{2} - X'\} + BSX'(\alpha+\beta)^{2}\{(1+\gamma)X' + 2Y'\}\}\right)^{-1}$$
(26)

The specific heat  $C_V$  is plotted as the function of volume V in figure 5(b) for three different values of k. The positivity of specific heat is obtained for all considered values of k. It should be noted that when temperature T is zero, the  $C_V$  vanishes, which assures the validity of third law of thermodynamics.

#### 4 Discussions

In this work, we have studied the thermodynamic properties of cosmological fluid described by the van der Waals equation of state in the framework of flat FRW universe. The phenomena of late time accelerated expansion of the universe is studied through different physical parameters like pressure, effective EoS, deceleration parameters and squared speed of sound. Figure 1 shows the positive and negative behavior of pressure P. We have also observed that at small volume, the universe is accelerating and it evolves to decelerated phase at large volume. From figure 2, we have observed that the EOS parameter  $\omega$  transits from -1 to positive values as volume increases. That means it yields cosmological constant model for small volume, then it generates the quintessence region and goes to positive region (tends to  $\gamma$ ). From figure 3, we have seen that the deceleration parameter q increases from -1 to the



**Figure 5:** Plots of  $\frac{\partial P}{\partial V}$  and  $C_V$  respectively against V for  $\gamma = 0.7, \beta = 1, \alpha = 20$  k = 5 (blue curve), k = 10 (red curve) and k = 20 (green curve).

positive value (tends to  $\frac{1+3\gamma}{2}$ ). It describes acceleration at small volumes whereas at large volume, it exhibits decelerating behavior. So, a transition from accelerating to decelerating universe occurred. That means the deceleration parameter q yields cosmological constant model for small volume and for large volume, it crosses the quintessence region. It shows  $q \to -1$  with decreasing V i.e it yields cosmological constant model as  $V \to 1$ . For the stability analysis of the model, the squared speed of sound  $V_s^2$  has drawn in figure 4. We have observed that for  $0 \le V \lesssim 9$ , graph shows  $V_s^2 > 0$  while for  $V \gtrsim 9$ , graph shows  $V_s^2 < 0$ . So the model is classically stable for small volume and for large volume, it shows unstable behavior. Finally, we have examined the thermodynamic stability of the considered fluid using adiabatic, isothermal and specific heat conditions. From figure 5(a), we have seen that  $\left(\frac{\partial P}{\partial V}\right)_S < 0$  at large volumes and at small volume, it is positive and tends to zero as  $V \to 0$ . So, the adiabatic condition is satisfied. From figure 5(b), we have observed that he specific heat  $C_V$  is always positive. So the third law of thermodynamics is obeyed for Van der Waals fluid. For all the figures, we have assumed the considered values of k (= 5, 10, 20). It should be noted that when temperature T is zero, the  $C_V$  vanishes, which assures the validity of third law of thermodynamics.

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